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# Hand's Baking Powder.

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## NOTES ON SCIENCE AND INDUSTRY.

The introduction of the Fauvel process into the Wyoming mining districts, by W. E. Meade of Cheyenne, has recently been announced. This process, according to an interesting description by Mr. Sanford of the Royal Mint, London, employs heat, air, and water to oxidize the refractory elements and liberate the gold, without the use of either chemicals or electricity.

By the time the ore has reached the bottom of the tank, the water has been cooled sufficiently ready for further treatment, this generally consists in wet amalgamation, and as was previously suggested it may be desirable to use a continuous system of treatment, in which the ore is continuously sprayed by a running stream of cold water. Acting on this, Foxwell made numerous experiments, and found that the best results were obtained when ore in a bath of cold water, the results of which are shown in the accompanying photograph, is allowed to fall into the water, causing an instantaneous generation of steam, accompanied by a violent explosion, and the ore is completely submerged, any glaze of film which may have formed on the surface of the ore is broken up and washed down to a remarkably fine state and rendered very brittle. Thus, exceedingly fine crushing is effected, and the surface of the ore is rendered greatly increased. The ore is reduced so minutely that it is blown away by the slightest breeze, and the film of oxide, so apt to coat the grains, is removed, the gold being rendered clean and shining. Formerly, after wet amalgamation, the same method of photographing designs on tissue is reported: The tissues are first placed in a bath of primuline, to which common salt has been added; they are then washed, and afterwards immersed for a quarter of an hour in a solution of potassium cyanide, and finally washed strongly acidulated by nitric or hydro-

cholic acid. After another washing, and while the tissue is still moist, or after it has been dried in the dark, the photographic negative is placed in contact with the tissue, and the whole exposed to the light, on a clear day and in the sun, for 10 to 15 minutes, or in the shade, suffering, but on a dull day thirty minutes being required. The tissue is then washed in water, and a small piece of tissue exposed being used as a test of the printing, the tissue is pinched in the dark, the development being carried out as ordinary photograph printing, familiar to all; the color is developed in this method at the discretion of the light, while it is not developed elsewhere, and interesting work is the result.

Wood pulp papers, now coming into vogue for the manufacture of the various kinds of printing process, the pulp, says the *Scientific American*, is agitated with water and rolled upon a tube, and, after being thus wound to a sufficient thickness around the tube and the extra removed, the paper is placed between two glass ends and the interior mould is withdrawn, leaving the wood pulp tube, which is held on suitably, and the water is allowed to run off and evaporated. The further process consists of

lifting it into view, the bottom of aspiration pipe is covered with a layer of mud, while the whole substance; the ends are then squared up and the threads cut, or, tapes finished with a sharp knife. The ends of the aspiration pipe. This material, when finished, possesses high electrical resistance, permitting it to be used in the manufacture of cables, and, as a non-conductor, it is free from being impaired by electrolytic action from the water in the pipes. The material is also free from the risk of impairing the water and gas pipes in cities where there is a great deal of water, and, as a non-conductor, it is free from being impaired by electrolytic action from the water in the pipes. Its resistance to acids and alkalis fits it for use in chemical works, and, as a non-conductor, it is free from being impaired by electrolytic action from the water in the pipes. The bursting strength of the pipe is 200 pounds per square inch, according to the size.

These facilities attending the production of chlorine and the impossibility of transmitting it in the gaseous state, remarks a writer in *Knowledge*, have combined to render chloride of lime in this country and chloride of soda in France the only forms in which this indispensable bleaching agent can be put upon the market.

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The ingenious Walraad-Licensed process for blowing iron is explained in detail in a paper recently read before the Iron and Steel Institute, London. The process consists simply in adding to the metal in the converter at the end of the ordinary blow a definite quantity of metallic ferrosilicon, then making the after-blow, turning the converter round, and blowing the metal burned out, and adding the ordinary final amount of ferro-manganese, &c., as required by the analysis. The process is described in the following words:—"The process are, first, that an ordinary Bessemer pig iron may be used with two to three per cent. silicon; secondly, the combustion of the additional silicon produces such a large amount of heat as to cause the metal to become more fluid; third, as the silicon burns out, it leaves a residue of silica, which, as the metal is being, therefore, poured, and with no gas cavities; then, too, the metal, being so fluid and so free from gas, is more easily treated with ferro-manganese or other substance added, such as aluminium, is more effective, and remains in the metal longer."

A method of testing the amount of moisture in

test has been devised by the Institution of Engineers and Shipbuilders, Scotland. The principle in this case, more particularly applicable to marine engines, consists in comparing the saltiness of the steam with that of the water in the boiler. The test, as explained, is a simple one, and the results of the reaction is so delicate that, with only one percent of salt in the boiler, one percent of priming will be obtained. To obtain the test to the second decimal, To one part of salt boiler water there is added 100 parts of distilled water. To this mixture there is poured a small quantity of concentrated solution of yellow chromate of potash; then, after shaking, the mixture is allowed to settle, and the liquid is poured into a test glass. The tenth per cent. of this salt is shown added, and the color of the salt water turns locally orange. The color of the water in the boiler to which all the salt has been added is the whole; this color changes once from pale yellow to orange, and then to red. The test is repeated, and then the experiment is repeated on the condensed steam from the engine supplied with the same water. The amount of priming is expressed in terms of the amount of salt in the water, and the amount of priming in per cent.